

FACILE SYNTHESSES OF 3-SPIRO- AND 3,6-BRIDGED 2,5-PIPERAZINEDIONES

Chung-gi Shin,* Yoshiaki Sato, and Juji Yoshimura[†]

Laboratory of Organic Chemistry, Kanagawa University, Kanagawa-ku, Yokohama 221

[†]Laboratory of Chemistry for Natural Products, Tokyo Institute of Technology,
Midori-ku, Yokohama 227, Japan

Summary: The intramolecular addition of hydroxyl group of 3-(3-hydroxyl)propylidene-2,5-piperazinedione to 3-position and the same substitution of 3-(3-hydroxy)propyl derivative to 6-position gave the corresponding 3-spiro- and 3,6-bridged 2,5-piperazinediones, respectively.

Recently, much attention is being denoted to the synthesis of fumitremorgins,¹⁾ verruculogen,²⁾ and bicyclomycin,³⁾ containing a hydroxyl group at 3-position and a bicyclic structure of 2,5-piperazinedione (PDO) derivatives. In connection with the synthesis of α -alkoxy- α -amino acids,⁴⁾ more recently, the convenient and facile addition of alcohols and water to the C=C bond of 3-alkylidene- and 3,6-dialkylidene-PDO has been presented by us.⁵⁾ On the other hand, concerning the preparation of 3-alkoxy-PDO, the similar addition reaction⁶⁾ and the another preparative method have been reported.^{7,8)} Here, we wish to report the more applicable intramolecular addition and substitution of 3-(3-hydroxy)propylidene- and -propyl-PDO, whose results are devoted to the suggestion for the synthesis of the bicyclic structural skeleton of bicyclomycin.

As the starting 3-alkylidene-PDO, ethyl 2-chloroacetyl-amino-4-ethoxycarbonyl-2-butenate [Bp 162-164 °C/2mmHg. IR: 3300 (NH), 1740 (COOEt), 1690 and 1520 (NHCO) cm⁻¹. NMR: δ 6.94, 7.36t (-CH=, J=7.0Hz), 8.28, 8.66 (NH)], derived by the condensation of ethyl 4-ethoxycarbonyl-2-oxobutanoate with chloroacet-

by chromatography on a silica gel column using a mixture of CHCl_3 -acetone (25 : 1 v/v) as the eluent to give two kinds of colorless crystals, which were identified as 3-(4-bromocyclopentyl-1-oxy)spiro-PDO [6a; δ 5.05t (-CHBr-, $J=7.5\text{Hz}$)] from first fraction and the diastereoisomer [6b; δ 4.22m (-CHBr-)] from the second, respectively. Subsequently, the bromine of each isomer thus obtained was debrominated with 10% Pd-C in the presence of Et_3N by the catalytic reduction to give the corresponding same spiro-PDO derivative [7; δ 2.02m (- CH_2 - at 4-position)] in a fairly good yield.

Furthermore, in order to synthesize the desired 3,6-bridged PDO derivative, successive reduction of 3 with 10% Pd-C to the corresponding saturated PDO [8; δ 4.02m (3-H)] and the methylation of 8 with methyl iodide to the corresponding N-methylated PDO [9; δ 4.03t (3-H, $J=5.5\text{Hz}$)] were carried out by the usual methods, followed by the removal of t-butyl group by CF_3COOH to give 1-benzyl-3-(3-hydroxy)propyl-4-methyl-PDO [10; 3350 (OH), 1660 (CO) cm^{-1} , δ 4.40 (OH), 4.04t (3-H, $J=5.0\text{Hz}$)] in an 83% overall yield from 3. Subsequently, a solution

Table 1. The yields, melting points, and spectral data of 4-9, and 10

Compound No.	Yield (%)	Mp $^{\circ}\text{C}$ ^{c)}	NMR spectrum, δ in CDCl_3			IR spectrum ^{f)} (cm^{-1}) NH
			NH	6-H (J_{Hz})		
<u>4</u>	96	138	8.54	3.94d (1.5)	3170	
<u>5</u>	95	symp	8.34	3.89q (18.0) ^{e)}	3220	
<u>6a</u>	59	142-143	7.92	3.91q (18.0) ^{e)}	3270	
<u>6b</u>	25	150-151	8.30	3.88d (4.0)	3180	
<u>7</u>	91 ^{a)} 90 ^{b)}	94-95	8.79	3.89q (18.0) ^{e)}	3200	
<u>8</u>	92	120-122	7.74	3.77s	3240	
<u>9</u>	98	97-98	—	3.83q (18.0) ^{e)}	—	
<u>10</u>	92	73-75 ^{d)}	—	3.86d (2.5)	—	

a) From 6a. b) From 6b. c) Colorless needles. d) Colorless prisms. e) AB quartet. f) In KBr.

of equimolar 10 (3.99 mmol) and NBS in CHCl_3 (10 ml) was refluxed for about 1 hour and then the reaction solution was washed twice with water. The organic layer was dried over anhydrous Na_2SO_4 and then concentrated to give colorless

crystals, which was identified as 8-benzyl-10-methyl-8,10-diaza-2-oxobicyclo-[4.2.2]decan-7,9-dione [11; yield 68%, mp 136-138 °C. IR: 1670 (CO) cm^{-1} . NMR: δ 4.12dd (3-H, $J=3.0\text{Hz}$ and $J=4.5\text{Hz}$), 5.12s (6-H)].

In consequence, it could be first generalised the inter- and intramolecular addition of hydroxyl group in the presence of NBS to 3-alkylidene-PDO and the intramolecular substitution to 6-position giving the corresponding 3-alkoxy-, 3-spiro-, and 3,6-bridged PDO.

From the above results, the formation mechanism of 4 and 6 is presumed that the N-bromo derivative of 3 and 2 is initially formed as an intermediate, followed by the 1,3-migration of the bromine to yield unstable intermediate containing C=N bond in ring, to which hydroxyl group immediately adds to give 4 and 6. In fact, when the nitrogen atom was protected with acyl group, the addition of hydroxyl group to C=C bond did not proceed.^{5,10} On the other hand, in the case of 10, it is supposed that bromine attacked initially to the ring methylene at 6-position of 10 to form 1-benzyl-6-bromo-4-methyl-3-(3-hydroxy)propyl-PDO as an unstable intermediate, in which the subsequent cyclization between the bromo and hydroxylpropyl groups by the nucleophilic substitution proceeds to give 11.

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